The Existing Substances Risk Assessment of Nickel was completed in 2008. The straightforward explanation of the goals of this exercise was to determine if the ongoing production and use of nickel in the EU caused risks to humans or the environment. The European Union launched the Existing Substances regulation in 2001 to comply with Council Regulation (EEC) 793/93. “Existing” substances were defined as chemical substances in use within the European Community before September 1981 and listed in the European Inventory of Existing Commercial Chemical Substances. Council Regulation (EEC) 793/93 provides a systematic framework for the evaluation of the risks of existing substances to human health and the environment.

The conceptual approach to conducting the environment section of the EU risk assessment of nickel included the following steps (Figure 1):

- Emissions of nickel and nickel compounds to the environment were quantified for the whole life cycle, i.e. from production, use, and disposal;
- Concentrations of nickel resulting from these emissions were determined in relevant environmental media (water, sediment, soil, tissue) at local and regional scales (PECs);
- Critical effects concentrations (PNECs) were determined for each of the relevant environmental media;
- Exposure concentrations were compared to critical effects concentrations for each of the relevant environmental media (risk characterization); and
- Appropriate corrective actions (also described as risk management) were identified for situations where exposure concentrations were greater than critical effects concentrations. Where exposure concentrations were below critical effects concentrations, there was no need for concern or action.

The initial EU Risk Assessments for Nickel and Nickel Compounds were developed over the period from 2002 to 2008 but the European Commission identified some remaining data gaps with respect to the sediment compartment (Official Journal of the European Union 2008). Therefore, a multilaboratory, multiphase research project was conducted to provide a scientific basis for a bioavailability based approach for assessing risks of nickel in sediments. The laboratory testing initiative was conducted in three phases to satisfy the following objectives: 1) evaluate various methods for spiking sediments with nickel to optimize the relevance of sediment nickel exposures; 2) generate reliable ecotoxicity data by conducting standardized chronic ecotoxicity tests using 10 benthic species in sediments with low and high nickel binding capacity; and 3) examine sediment bioavailability relationships by conducting chronic ecotoxicity testing in sediments that showed broad ranges of acid volatile sulfides, organic carbon, and iron. A subset of 6 nickel-spiked sediments was deployed in the field to examine benthic colonization and community effects. The sediment testing program yielded a broad, high quality data set that was used to develop a Species Sensitivity Distribution (SSD) for benthic organisms in various sediment types, a reasonable worst case predicted no-effect concentration for nickel in sediment (PNECsediment), and predictive models for bioavailability and toxicity of nickel in freshwater sediments (Schlekat et al., 2016).

1 INTRODUCTION

Environmental risks are typically characterized in the risk assessment framework by comparing exposure concentrations and critical effect concentrations. In Organisation for Economic Co-operation and Development (OECD) countries, critical effect concentrations for metals are based on Predicted No Effect Concentrations (PNEC), which are typically derived from long-term laboratory-spiked ecotoxicity tests performed with highly soluble, almost completely dissociated metal salts in artificial laboratory standard sediments or natural sediments. It has been recognized that results from these whole sediment tests may be influenced by several parameters (e.g., sediment composition, spiking method, feeding mode, bioavailability) and that caution is needed in selecting and developing the appropriate test methods. For example, traditional sediment spiking methods that involve adding soluble nickel metal salts to sediments without further pH amendment result in significant diffusion of nickel from the sediment compartment to the water compartment (Vandeghuchte et al., 2007). During 2011-2015, a multi-laboratory, multiphase research project was conducted to address this issue (i.e. unrealistic high exposure conditions) and to provide a scientific basis to derive a bioavailability based approach for assessing risks of nickel (Ni) in sediments (Schlekat et al., 2016). Besides developing new spiking procedures (Brumbaugh et al., 2013), the nickel sediment research program demonstrated that...
the chronic toxicity of nickel in sediments was influenced by several physicochemical characteristics of the tested sediments, with the highest toxicity found in sediments with low Acid Volatile Sulfides (AVS) concentrations, low Total Organic Carbon (TOC), low total recoverable iron (Fe), and low Cation Exchange Capacity (CEC) (Besser et al., 2013; Vangheluwe et al., 2013; Vangheluwe and Nguyen, 2015). Therefore, nickel toxicity can vary considerably among sediments with different physico-chemical characteristics and tests with the same sediment species that are performed using different sediment types can produce different results. Consequently, bioavailability models were developed to directly compare sediment toxicity and to generate sediment threshold values, e.g., PNECsed.

This fact sheet provides an overview of the nickel sediment bioavailability models and demonstrates how this information can be used to estimate site-specific bioavailable nickel PNEC values. The availability of sediment physico-chemistry data, such as AVS content, allows site-specific nickel PNEC values to be calculated and a more accurate site-specific risk characterization to be conducted.

## 2 NICKEL BIOAVAILABILITY MODELS

Bioavailability and chronic toxicity of nickel to sediment organisms varies based on sediment characteristics. To make comparisons between laboratory toxicity data, results must be normalized to a common set of conditions using bioavailability models. Bioavailability models can be used to derive site-specific HC5/PNEC values for sites in which appropriate sediment properties have been quantified. For nickel, chronic sediment toxicity tests are available for 10 species of sediment dwelling organisms conducted in nickel-spiked sediments representing sediments with low and high nickel binding capacity (i.e., low AVS/Low TOC and high AVS/high TOC) (See Fact Sheet 8). In addition, chronic toxicity tests were conducted with several additional nickel-spiked sediments with a wide range of AVS and TOC concentrations to characterize relationships for 7 test species between nickel toxicity and sediment characteristics (i.e., bioavailability regression models) (Besser et al., 2013; Vangheluwe and Nguyen, 2015; Vangheluwe et al., 2013). The chronic regression models for nickel were developed/calibrated based on sediments that represent the full range in physico-chemical parameters (AVS, OC) representative for the EU. EC20 values, expressed as either total recoverable nickel or SEMnicks, showed significant relationships with a range of sediment parameters, including AVS, total recoverable iron, TOC, CEC, silt, total recoverable manganese, and SEMmang. The importance of sediment phases other than AVS indicates that the relationships should be relevant for oxic sediments as well as anoxic sediments. For all species tested, the sediment parameter showing the strongest linear relationship was AVS. AVS has already been demonstrated as being one of the predominant factors controlling toxicity of divalent metals (Di Toro et al., 1992; Ankley et al., 1991 and 1996). An overview of the slopes and intercepts of all significant regression models relating the toxicity of nickel ([Ni] in mg/kg dry wt.) to AVS in sediments is presented in Table 1.

<table>
<thead>
<tr>
<th>Species</th>
<th>Life Strategy</th>
<th>Intercept</th>
<th>Slope</th>
<th>Example Picture</th>
</tr>
</thead>
<tbody>
<tr>
<td>H. azteca</td>
<td>Swimmer, sprawler, surface deposit feeder</td>
<td>2.65</td>
<td>0.492</td>
<td></td>
</tr>
<tr>
<td>S. corneum</td>
<td>Burrower, surface deposit feeder</td>
<td>2.73</td>
<td>0.478</td>
<td></td>
</tr>
<tr>
<td>G. pseudolimnaeus</td>
<td>Swimmer, sprawler, surface deposit feeder</td>
<td>2.8</td>
<td>0.358</td>
<td></td>
</tr>
<tr>
<td>E. virgo</td>
<td>Burrower, surface and subsurface feeder</td>
<td>2.21</td>
<td>0.218</td>
<td></td>
</tr>
<tr>
<td>C. riparius</td>
<td>Burrower, surface and subsurface feeder</td>
<td>2.85</td>
<td>0.180</td>
<td></td>
</tr>
<tr>
<td>Hexagenia sp.</td>
<td>Burrower, surface and subsurface feeder</td>
<td>2.35</td>
<td>0.175</td>
<td></td>
</tr>
<tr>
<td>T. tubifex</td>
<td>Burrower, subsurface feeder</td>
<td>3.05</td>
<td>0.125</td>
<td></td>
</tr>
</tbody>
</table>

Table 1: Overview slopes and intercepts of all significant regression models relating the toxicity of nickel ([Ni] in mg/kg dry wt.) to AVS in sediments.
Although the effect of decreasing toxicity with increasing AVS was consistently observed for all species, the magnitude of the effect was not similar among species, and these differences appear to be linked with organism behavior. The strongest mitigating effects of AVS are observed for those species with an epibenthic lifestyle such as *H. azteca*, *S. corneum*, and *G. pseudoflavimanae*, with slopes ranging from 0.358 to 0.492 (Table 1). The relationships are less pronounced for the benthic species *T. tubifex*, *C. riparius*, *Hexagenia sp.*, and *E. virgo* that exhibit more burrowing activity and subsurface feeding (i.e. slopes ranging from 0.125-0.22).

3 APPLICATION OF BIOAVAILABILITY MODELS

The empirically based regression models predicting nickel toxicity in spiked sediments based on sediment property parameters can be used to normalize the available toxicity data set to specific bioavailability conditions. For example, a Realistic Worst Case (RWC) PNEC can be derived by recalculating the nickel sediment toxicity database toward a reference situation, such as 10th percentile of the regional AVS/ distribution, in case no actual or historical AVS data are available. Alternatively, a PNEC can be derived for the actual AVS concentrations occurring at the site as follows:

1) Link the NOEC/ECx values of the chronic ecotoxicity database (as total metal concentrations) with the relevant sediment parameters of the sediment (e.g., AVS) in which the test was performed.

2) For the regression models [taking the form \( \log(\text{ECx}) = \text{intercept} + \text{slope} \times \log(\text{abiotic factor}) \)], the corresponding organisms specific slopes can be used to normalize the NOEC/ECX values to “reasonable worst case” sediment properties (e.g., 10th percentile AVS) or to specific local/regional conditions (actual or historical AVS concentrations prevailing on the site under investigation). The normalization equations for RWC and site-specific conditions are given below in Equations 1 and 2.

\[
\text{Equation 1—RWC}
\]

\[
\text{EC}_{x,\text{RWC}} = \frac{\text{EC}_{x,\text{test}} \times \text{abiotic factor}_{\text{RWC}}}{\text{abiotic factor}_{\text{test}}} \]^{\text{slope}}
\]

RWC = realistic worst case scenario corresponding to a maximized bioavailability

\( \text{test} = \text{scenario with typical local or regional conditions for which the EC}_{x,\text{test}} \text{ is derived} \)

The different bioavailability models have been used to normalize the final toxicity dataset to nine different bioavailability scenarios (eight specific sediments ranging from low to high AVS content and one hypothetical RWC condition). The different sediments are described in the paper of Vangheluwe et al. (2013). Bioavailability models are available for seven species. For all the derived ECx values the oligochaete *L. variegatus* is the only test organism without a specific bioavailability model, and hence, this is the only data point in the Species Sensitivity Distribution (SSD) for which the bioavailability model for another species needs to be used. The choice of the model to use for *L. variegatus* was based on biological and practical considerations, including similarities in life history and behavior with other species for which bioavailability models were established, and the degree to which different bioavailability models reduced observed intra-species variability for *L. variegatus*. Oligochaetes such as *L. variegatus* alter their immediate environment through the formation of I-shaped burrows. Tubificid worms feed on deposits on some depth. In contrast with *Hexagenia*, the micro-habitats of oligochaetes are not irrigated with oxygenated water; nevertheless, this behavior will minimize the mitigating capacity of a bioavailability factor such as AVS as reflected in the smaller slope of the AVS model developed for *Hexagenia* and *T. tubifex*. Because of the close similarity between tubificid/oligochaete worm behavior the *T. tubifex* model is used to normalize the *L. variegatus* data. This choice is considered precautionary as it has the smallest slope of all species tested (i.e. 0.125). Species to species “read across,” however, may introduce additional uncertainty. Figure 2 represents the lognormal functions normalized for AVS that were fitted through the eight data points for eight sediments and the RWC. All functions were accepted at \( P < 0.05 \).

### Table 2: Overview of the water chemistry and median HC/PNEC values for the different selected EU eco-regions (values between brackets are 90% confidence intervals)

<table>
<thead>
<tr>
<th>Eco-Region</th>
<th>Sediment Chemistry (µmol AVS/g dry wt.)</th>
<th>Median HC (mg nickel kg dry wt.)</th>
<th>PNEC (mg nickel kg dry wt.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Generic Reasonable Worst Case Sediment</td>
<td>0.8</td>
<td>109 (40-182)</td>
<td>109</td>
</tr>
<tr>
<td>Spring River, Missouri, USA</td>
<td>0.9</td>
<td>115 (43-191)</td>
<td>115</td>
</tr>
<tr>
<td>Dow Creek, Michigan, USA</td>
<td>1.0</td>
<td>121 (46-201)</td>
<td>121</td>
</tr>
<tr>
<td>Brakel 1, Belgium</td>
<td>2.6</td>
<td>165 (66-264)</td>
<td>165</td>
</tr>
<tr>
<td>St. Joseph River, Michigan, USA</td>
<td>3.8</td>
<td>185 (75-296)</td>
<td>185</td>
</tr>
<tr>
<td>Raisin River (site 2), Michigan, USA</td>
<td>6.1</td>
<td>210 (85-337)</td>
<td>210</td>
</tr>
<tr>
<td>Brakel 2, Belgium</td>
<td>6.2</td>
<td>212 (86-339)</td>
<td>212</td>
</tr>
<tr>
<td>Raisin River (site 3), Michigan, USA</td>
<td>8.0</td>
<td>225 (91-336)</td>
<td>225</td>
</tr>
<tr>
<td>US Geological Survey Pond 30, Missouri, USA</td>
<td>12.4</td>
<td>249 (99-403)</td>
<td>249</td>
</tr>
<tr>
<td>Lampernisse, Belgium</td>
<td>24.5</td>
<td>284 (108-469)</td>
<td>284</td>
</tr>
<tr>
<td>South Tributary Mill Creek, Michigan, USA</td>
<td>24.7</td>
<td>284 (108-470)</td>
<td>284</td>
</tr>
<tr>
<td>West Bearskin Lake, Minnesota, USA</td>
<td>38.4</td>
<td>305 (111-515)</td>
<td>305</td>
</tr>
</tbody>
</table>
Figure 2: Cumulative frequency distributions of the EC10 values (n = 7) (expressed as mg Ni/kg dry wt.) from the nickel chronic toxicity tests towards sediment-dwelling organisms, normalized towards prevailing AVS conditions. Observed data and log-normal curve for the dataset fitted on the data.

A summary of the estimated HC5.50 values (with the 5-95% confidence limits) for the different log-normal distributions is provided in Table 2. The HC5.50 values obtained for the different bioavailability scenarios range with the AVS model from 109-305 mg/kg dry wt. The PNEC was calculated as the HC5-50 multiplied by an AF (assessment factor) of 1.

4 CONSIDERATIONS FOR APPLICATION

4.1 AVS DATABASES

To apply an AVS normalization, a good understanding of the amount and geographical distribution of AVS in sediments is needed. AVS concentrations are not yet measured on a routine basis in sediment monitoring programs but some data have been collected on the AVS distribution in different countries within the context of metal risk assessments where the SEM-AVS concept has been used. An overview of AVS ranges that are typically encountered in sediment is given per country in Figure 3 as Box-Whisker plots. A box and whisker diagram, or boxplot, provides a graphical summary of a set of data based on the quartiles of that data set. The ‘box,’ or rectangle, in Figure 3 contains 50% of the data, and the extremes of that box are the 25th percentile and 75th percentile. Each ‘whisker’ represents the remaining 25% of the data and the extremities of these whiskers are the minimum and maximum values of the data.

The largest database available is for the Flanders region of Belgium. This database, which is representative for EU low midland rivers, contains 200 sediments sampled over a depth of 0-10 cm. The 50th percentile of the AVS distribution yields an AVS value of 8.7 µmol/g dry wt. (Vangheluwe et al., 2005). The lowest AVS concentration in the Flanders dataset is 0.045 µmol/g dry wt. The 10th percentile is 0.77 µmol/g dry wt. The latter value has been used in the different ongoing metal risk assessments to be used as a generic default correction value for low midland rivers (the Netherlands, Germany, and possibly Northern France) when site-specific measurements are lacking.

In 2008, three additional countries were sampled: Finland, United Kingdom, and Spain (Vangheluwe et al., 2008). Although the intention was to sample in the spring season (April-May) when AVS concentrations are expected to be the lowest, this was only possible for Finland. In the United Kingdom, sampling was conducted in June-September and Spain in October. In Finland, a total of 25 samples were taken (13 lakes, 12 rivers). Analysis of the AVS concentrations gives a 10th percentile of 1 µmol/g dry wt. and a 50th percentile of 11 µmol/g dry wt. The lowest concentration measured was 0.3 µmol/g dry wt. For the United Kingdom, 16 sediments from 16 different rivers were sampled. Analysis of the AVS concentrations gives a 10th percentile of 0.31 µmol/g dry wt. and a 50th percentile of 7.95 µmol/g dry wt. The lowest concentration measured was 0.071 µmol/g dry wt. For Spain, 20 samples of the river Ebro were sampled. Analysis of the AVS concentrations gives a 10th percentile of 3.68 µmol/g dry wt. and a 50th percentile of 13.5 µmol/g dry wt. The lowest AVS concentration measured was 1.7 µmol/g dry wt.

Burton et al. (2007) investigated AVS concentrations for 84 sites in wadable streams of 10 countries and nine ecoregions of Europe. The results showed AVS concentrations ranging from 0.004 µmol/g dry wt. to 44 µmol/g dry wt. with a median value of 0.1 µmol/g dry wt. (sample depth 0-5 cm) and an average value of 2.5 µmol/g dry wt. It should be noted that sediments in this program were collected in head streams resulting in very low AVS and SEM levels.
Figure 3: Overview Box-Whisker plots AVS data (µmol/g dry wt.)
For Belgium (Flanders, n = 202 stations), the Netherlands (n = 29 stations), Hungary (n = 9 stations), United Kingdom (n = 16 stations), Finland (n = 25 stations), Spain (n = 20 stations), Serbia (n = 12 stations) For Italy (n = 4 stations) and Sweden (n = 4 stations) only the individual sampling points are given

4.2 SEDIMENT SAMPLING RECOMMENDATIONS

The use of AVS is sometimes criticized because of the dynamic behavior of AVS in natural systems. AVS concentrations have shown temporal and spatial (horizontal and vertical) variations depending on sediment type and hydrological conditions (Poot et al., 2007). Most often the AVS concentration increases with increasing sediment depth (even over small sample distances 0-10 cm) and is linked to the redox gradient present in the sediment (Van Den Berg et al., 1998; Van Den Berg et al., 2001a, 2001b). In addition, there seems to be a strong seasonal component where AVS concentrations tend to be the higher at the end of the summer and during fall and lower in winter and spring (Howard and Evans, 1993; Van den Hoop et al., 1997; Grabowski et al., 2001). However, it should be noted that transient nature of AVS in this regard maybe overstated as most of the studies relate to uncontaminated sediments where the oxidation of iron sulfide in sediments cannot be taken as indicative of the oxidation of the other metal sulfide complexes, which are more stable [e.g., copper, zinc, cadmium sulfide (Peterson, 1996) and nickel sulfide (Buykx et al., 2000)].

Despite the higher stability of nickel sulfide complexes, both temporal and spatial variations are important to be considered when collecting SEM-AVS data. As AVS concentrations have the tendency to be lower in spring and winter than in summer, it is recommended to sample sediments late winter/early spring. Furthermore, it is recommended to sample the top layer (0-5 cm) as AVS concentrations are lower than those found in the deeper layers (>10 cm) (Van Den Berg et al., 1998; De Lange et al., 2008). As such, realistic worst case exposure conditions are guaranteed to conduct bioassays.

4.3 ANALYTICAL MEASUREMENTS

Acid volatile sulfide is defined operationally as those sulfides that are readily extracted by the cold extraction of sediment in approximately 1 molar HCl acid (Allen et al., 1993). Another term that is used in conjunction with AVS is SEM. SEM (Simultaneously Extracted Metal) can be defined as the metal, which is simultaneously extracted under the same conditions under which the AVS content is determined. AVS is a complex and variable fraction represented by a variety of reduced sulfur components, although often dominated by relatively labile iron and manganese monosulfides (Morse, 2004). In the past multiple techniques, employing different chemical reagents and methodologies have been used to extract AVS from sediment. As the efficiency of sulfide extraction from minerals including pyrite varies amongst methods different results can be expected from these measurements. In part to promote a greater intercomparability of AVS and SEM results extraction with 1 molar HCl has been proposed (Allen, 1991). Nevertheless, it has been shown by Hammerschmidt and Burton (2010) that caution is needed to interpret results that lack well described quality control procedures which may lead to a high variability among laboratories. This was even more apparent in cases where sediments with low AVS concentrations (< 0.5 µmol/g dry wt.) were used. However, recently Brumbaugh et al. (2011) has shown, in an interlaboratory comparison of measurements of Acid Volatile Sulfide and Simultaneously extracted nickel in spiked sediments, that measurements of AVS and SEM-AVS can be reproducible among different laboratories when performed with structured analytical guidelines.
APPLICATION OF THE SEM-AVS CONCEPT IN A RISK ASSESSMENT CONTEXT

Figure 4 represents the tiered approach that can be used to refine the risk assessment using the SEM-AVS concept.

No risk is identified when the total nickel concentration is smaller than the RWC PNEC. This PNEC for nickel can be derived using sediments high in bioavailability (see Fact Sheet 8).

If a risk is identified and historical AVS data are available for the receiving water or river/lake systems similar in characteristics these data can be used to calculate the PNEC AVS normalized using the available nickel bioavailability models.

The actual risk can be calculated using actual measured SEM-AVS concentrations for the site under investigation.

In case SEM-AVS > 0 a further weight of analysis can be conducted (e.g., direct toxicity testing, macro-invertebrate analysis etc.).

Figure 4: Overview of tiered approach for assessing risks of nickel in sediments on a bioavailable basis applicable for a local scenario.

5 APPLICATION OF THE SEM-AVS CONCEPT IN A RISK ASSESSMENT CONTEXT

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Figure 4: Overview of tiered approach for assessing risks of nickel in sediments on a bioavailable basis applicable for a local scenario.
6 CONCLUSIONS AND NEXT STEPS IN RISK ASSESSMENT

When considering nickel toxicity in sediments, it is important to account for sediment bioavailability. Bioavailability models based on AVS binding have subsequently been developed to normalize nickel ecotoxicity data to a common site condition when deriving HC5/PNECs or sediment standards. This fact sheet presents the background information on the available sediment nickel bioavailability models and demonstrates how this information can be used to estimate site-specific bioavailable PNEC values. The availability of sediment physico-chemistry data, such as AVS content can be limited in some cases but sampling analyzing AVS concentrations can be done in a reproducible way allowing site-specific PNEC values to be calculated and a more accurate site-specific risk characterization to be conducted. However, the sole use of the SEM-AVS concept for nickel is conservative as it has been demonstrated that even in the absence of AVS, as is the case in oxic sediments, other partitioning phases such as binding to iron and manganese oxides may mitigate nickel toxicity in sediments (Costello et al., 2011).

7 LINKS TO NICKEL EU RISK ASSESSMENT DOCUMENTS

The final report on the environmental risk assessment of nickel and nickel compounds can be retrieved from the following website: http://echa.europa.eu/documents/10162/ceda88bc-2952-4c11-885f-542aac769bb3 (last accessed April 2018)

8 REFERENCES


Fact Sheets on the European Union Environmental Risk Assessment of Nickel

This is the ninth in a series of fact sheets addressing issues specific to the environment section of the European Union’s Existing Substances Risk Assessment of Nickel (EU RA). The fact sheets are intended to assist the reader in understanding the complex environmental issues and concepts presented in the EU RA by summarizing key technical information and providing guidance for implementation.

NiPERA welcomes questions about the concepts and approaches implemented in the EU RA. For inquiries, please contact:

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